The rates of disappearance of 1 were determined by measuring the decrease in the integrated area of the methylene doublet for the substrate (5.08 ppm) relative to the total integrated area for the methylene signals involved. All reactions were followed to 61-81% conversion. The values of k_1 and k_2 were calculated as slopes of the plots of the values of $\ln (a_0/a_0 - x_t)$ or $x_t/a_0 (a_0 - x_t)$ x_t) vs time.

Quantitatively meaningful integrals of ¹³C signals were obtained by using 40° pulses, followed with a 6-8-s pulse delay, during which time the BB decoupler was gated off. In this way all ¹³C signals except C-2 of the pyridine ring, yielded quantitative integrals to within $\pm 5\%$. Typically 10³ transients were collected, taking <1.8 h.

Acknowledgment. Financial assistance from the University of Cape Town and from the C.S.I.R. is gratefully acknowledged. S.C. thanks Beit Trust for a scholarship.

Registry No. 1, 99668-67-4; 2, 99668-68-5; (MeO)₂P(o)Cl, 813-77-4; sodium 2-pyridylmethoxide, 99668-69-6.

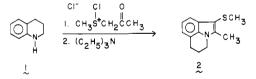
The Conversion of Tetrahydroquinoline into **Derivatives** of 8H-Pyrido[3,2,1-jk][1,3]benzodiazepines via [2,3]-Sigmatropic Rearrangements

Paul G. Gassman* and Sung Jai Lee

Department of Chemistry, University of Minnesota, Minneapolis, Minnesota 55455

Received September 3, 1985

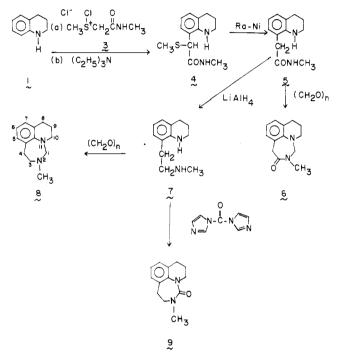
Although pyridobenzodiazepines have been previously reported in the literature,¹ the pyrido[3,2,1-jk][1,3]benzodiazepine ring system does not appear to be known. Recently, we described the use of tetrahydroquinoline as a starting material for the synthesis of derivatives of 4Hpyrrolo[3,2,1-ij]quinoline.² This transformation, which is exemplified by the conversion of 1 into 2, utilized the general concept of [2,3]-sigmatropic rearrangements of vlides derived from aza sulfonium salts.³ These rear-



rangements, which we have exploited in the synthesis of

indoles⁴ and oxindoles,⁵ are widely applicable to a variety of ring systems bearing both electron-donating and electron-withdrawing substituents. Thus, it seemed reasonable that we should be able to apply this same methodology in the synthesis of the 8H-pyrido[3,2,1-jk][1,3]benzodiazepines, a class of compounds which is attractive because of its close relationship to pharmaceutically intriguing structures.¹

Treatment of 2-(methylthio)-N-methylacetamide⁶ with 0.95 equiv of chlorine at -78 °C gave the chlorosulfonium chloride salt 3. Dropwise addition of 1 to a methylene chloride solution of 3, followed by ylide formation using triethylamine as base, afforded a 71% yield of 4. Raney



nickel reduction of 4 produced 5 in 85% yield. When a benzene solution of 5 was refluxed with paraformaldehyde for 12 h, the cyclization product, 2-methyl-1,2,9,10-tetrahydro-8H-pyrido[3,2,1-jk][1,3]benzodiazepin-3(4H)-one (6) was obtained in 76% yield.

Reduction of 5 with lithium aluminum hydride gave an 84% yield (71% based on 4) of 7 as an oil. When 7 was refluxed in benzene with 1 equiv of paraformaldehyde, a 91% yield of 1,2,3,4,9,10-hexahydro-2-methyl-8H-pyrido-[3,2,1-jk][1,3]benzodiazepine (8) was isolated. Refluxing 7 in tetrahydrofuran with 1 equiv of 1,1'-carbonyldiimidazole produced a 56% yield of the cyclic urea 9.

The investigation outlined above makes the 8Hpyrido[3,2,1-jk][1,3]benzodiazepines readily available for the first time.

Experimental Section

 α -(Methylthio)- α -(1,2,3,4-tetrahydroquinolin-8-yl)-Nmethylacetamide (4). Chlorine (4.0 mL, 88 mmol) was con-

⁽¹⁾ For selected examples, see: Ziegler, E.; Noelken, E.; Junek, H. (1) For selected examples, see: Liegier, E., Iverkein, E., ounen, I., Monatsh. 1962, 93, 708. Härter, H. P.; Liisberg, S. Acta Chem. Scand. 1968, 22, 3332. Welstead, W. J., Jr.; Chen, Y.-H. Ger. Offen. 2117 116; Chem. Abstr. 1972, 76, 59669x. Hester, J. B., Jr. U.S. Patent 3 714 149; Chem. Abstr. 1973, 78, 111380d. Gatta, F.; Landi-Vittory, R.; Tomassetti, Chem. Abstr. 1973, 77, 11380d. Gatta, F.; Landi-Vittory, R.; Tomassetti, Chem. Abstr. 1972, 76, 2010 and 2010 an Chem. Abstr. 1973, 78, 111380d. Gatta, F.; Landi-Vittory, R.; Tomassetti, M.; Nunez Barrios, G. Eur. J. Med. Chem.-Chim. Ther. 1972, 7, 480. Blickenstaff, R. T.; Wells, N. Org. Prep. Proc. Int. 1974, 6, 197. Hester, J. B., Jr. U.S. Patent 3793328; Chem. Abstr. 1974, 80, 957639. Gatta, F.; Tomassetti, M.; Zaccari, V.; Landi-Vittory, R. Eur. J. Med. Chem.. Chem. Ther. 1974, 9, 133. Merchant, J. R.; Chothia, D. S. Indian J. Chem. 1975, 13, 814. Kaemmerer, F. J.; Perrey, K. Ger. Offen. 2443567; Chem. Abstr. 1976, 85, 5700t. Horng, Y. S.; Hwang, C. C.; Liao, C. C.; Liu, C. S.; Sheu, C. F.; Yu, Y. H. J. Chin. Chem. Soc. (Taipei) 1980, 27, 55.

Gassman, P. G.; Roos, J. J.; Lee, S. J. J. Org. Chem. 1984, 49, 717.
 Claus, P.; Vycudilik, W. Monatsh. Chem. 1970, 101, 396. Gassman, Gruetzmacher, G. D.; Smith, R. H. Tetrahedron Lett. 1972, 497.
 Gassman, P. G.; Gruetzmacher, G. D. J. Am. Chem. Soc. 1973, 95, 588.
 Gassman, P. G.; Huang, C. T. Ibid. 1973, 95, 4453.
 Gassman, P. G.; Pruetzmacher, G. D. J. Am. Chem. Soc. 1973, 95, 588. Tetrahedron Lett. 1977, 2055.

⁽⁴⁾ Gassman, P. G.; van Bergen, T. J. J. Am. Chem. Soc. 1973, 95, 590, 591. Gassman, P. G.; Gilbert, D. P.; van Bergen, T. J. J. Chem. Soc., Chem. Commun. 1974, 201. Gassman, P. G.; van Bergen, T. J.; Gilbert, D. P.; Cue, B. W., Jr. J. Am. Chem. Soc. 1974, 96, 5495. Gassman, P. G.; Schenk, W. N. J. Org. Chem. 1977, 42, 3240.
(5) Gassman, P. G.; van Bergen, T. J. J. Am. Chem. Soc. 1973, 95, 2718. Gassman, P. G.; van Bergen, T. J.; Gruetzmacher, G. D. Ibid. 1973, 95, 656 6568. Gassmap, P. G.; van Bergen, T. J. J. Am. Chem. Soc. 1973, 95, 2718. Gassman, P. G.; van Bergen, T. J. Hidd 1974. 96, 5508. Gassman, P. G.; van Bergen, T. J. Gruetzmacher, G. D. Ibid. 1973, 95, 956 6568.

 ^{2:10.} Gassman, P. G.; van Bergen, 1. J.; Gruetzmacner, G. D. 101. 1973, 95, 6508. Gassman, P. G.; Gilbert, D. P.; Luh, T.-Y. J. Org. Chem. 1977, 42, 1340.
 (6) Mndzhoyan, A. L.; Afrikyan, V. G.; Badalyan, V. E.; Markaryan, E. A.; Khorenyan, G. A. Dokl.—Akad. Nauk Arm. SSR 1958, 27, 161; Chem. 4bcts. 1965. 54, 1585. Chem. Abstr. 1960, 54, 1525c.

densed into a dry ice-acetone-cooled (jacketed) addition funnel at -78 °C. The chlorine was added dropwise to 250 mL of dry methylene chloride at -78 °C. To the resultant pale yellow solution, under a static nitrogen pressure, was added 11.0 g (92 mmol) of 2-(methylthio)-N-methylacetamide⁶ in 10 mL of dry methylene chloride. A white precipitate formed. After the reaction mixture had been stirred for 20 min, a solution of 23.44 g (176 mmol) of 1,2,3,4-tetrahydroquinoline (1) in 25 mL of dry methylene chloride was added dropwise. The reaction mixture was stirred for 3 h at -70 °C, followed by addition of 40 mL of triethylamine. After being stirred for an additional hour at -70 °C, the reaction mixture was allowed to warm to ambient temperature (overnight). The organic mixture was washed with aqueous sodium carbonate solution, dried over anhydrous sodium carbonate, filtered, and concentrated to give a solid. This solid was recrystallized from ether-hexane to give 15.7 g (62.8 mmol, 71%) of white crystalline product: mp 137-139 °C; IR (CDCl₃) 3380 (w), 2940 (w), 2850 (vw), 1670 (s), 1610 (w), 1520 (m), 1450 (w), 1410 (w), 1380 (vw), 1310 (m), 1190 (w), 1115 (w) cm⁻¹; ¹H NMR (CDCl₃) δ 1.11-2.13 (m, 2 H), 2.03 (s, 3 H), 2.57-2.92 (m, 5 H), 3.16-3.48 (m, 2 H), 4.47 (s, 1 H), 5.03 (br s, 1 H), 6.36-7.10 (complex m, 3 H); ¹³C NMR (CDCl₃) δ 15.27 (q), 21.40 (t), 26.44 (q), 27.44 (t), 41.95 (t), 53.16 (d), 116.18 (d), 119.75 (s), 122.69 (s), 125.89 (d), 129.14 (d), 143.02 (s), 170.66 (s); exact mass, m/e 250.1143 (calcd for $C_{13}H_{18}N_2OS$, m/e 250.1139).

Anal. Calcd for $C_{13}H_{18}N_2OS$: C, 62.37; H, 7.25; N, 11.19. Found: C, 62.43; H, 7.27; N, 11.16.

2-(1,2,3,4-Tetrahydroquinolin-8-yl)-N-methylacetamide (5). To 7.5 g (30 mmol) of α -(methylthio)- α -(1,2,3,4-tetrahydroquinolin-8-yl)-N-methylacetamide (4) in a mixture of 100 mL of acetone and 100 mL of denatured ethanol was added 10 level teaspoons (approximately 30 g) of activated Raney nickel, and the reaction mixture was stirred for 30 min. The reaction mixture was filtered through a fritted glass funnel. The spent catalyst was washed with 50 mL of acetone. The combined filtrates were concentrated in vacuo. The resulting residue was taken up in 100 mL of methylene chloride and dried over anhydrous magnesium sulfate. Filtration and concentration gave 5.8 g (28.2 mmol, 94%) of product. Distillation gave 5.2 g (25.5 mmol, 85%) of pure 5: bp 175-190 °C (0.2 mm); IR (CDCl₃) 3420 (w), 3310 (vw), 2930 (w), 2850 (w), 1670 (s), 1610 (w), 1530 (m), 1415 (w), 1310 (w), 1280 (w) cm⁻¹; ¹H NMR (CDCl₃) δ 1.69-2.12 (m, 2 H), 2.58-2.90 (m, 5 H), 3.34 (s, 2 H), 3.18-3.47 (m, 2 H), 4.78 (br s, 1 H), 6.09 (br s, 1 H), 6.35-7.06 (complex m, 3 H); ¹³C NMR (CDCl₃) δ 21.47 (t), 26.11 (q), 27.2 (t), 40.30 (t), 41.83 (t), 116.13 (d), 118.85 (s), 122.01 (s), 128.26 (d), 128.61 (d), 143.39 (s), 171.91 (s); exact mass, m/e 204.1260 (calcd for C₁₂H₁₆N₂O, m/e 204.1262).

Anal. Calcd for C₁₂H₁₆N₂O: Č, 70.56; H, 7.90; N, 13.72. Found: C, 70.58; H, 7.87; N, 13.58.

2-Methyl-1,2,9,10-tetrahydro-8H-pyrido[3,2,1-jk][1,3]benzodiazepin-3(4H)-one (6). To 2.04 g (10 mmol) of 2-(1,2,3,4-tetrahydroquinolin-8-yl)-N-methylacetamide (5) in 25 mL of benzene was added 0.30 g (10 mmol) of paraformaldehyde, and the reaction mixture was refluxed overnight. The reaction mixture was concentrated to give a syrup, which was distilled to give 2.01 g (9.3 mmol, 93%) [bp 155-170 °C (0.2 mm)] of a syrup that gradually crystallized. Recrystallization from ether-hexane gave 1.65 g (7.6 mmol, 76%) of 6 as crystals: mp 92.0-93.5 °C; IR (CDCl₃) 2950 (w), 1670 (s), 1600 (w), 1480 (m), 1460 (m), 1400 (w), 1370 (w), 1330 (w), 1270 (w), 1210 (w), 1180 (w), 1110 (w) cm⁻¹; ¹H NMR (CDCl₃) δ 1.78 (quintet, 2 H), 2.63 (t, 2 H), 3.06 (s, 3 H), 3.78 (t, 2 H), 3.93 (s, 2 H), 4.73 (s, 2 H), 6.40-6.98 (complex m, 3 H); ¹³C NMR (CDCl₃) δ 23.28 (t), 27.27 (t), 34.37 (q), 43.29 (t), 50.35 (t), 67.41 (t), 117.79 (s), 118.59 (d), 127.19 (s), 127.42 (d), 130.00 (d), 143.04 (s), 171.94 (s); exact mass, m/e 216.1265 (calcd for $C_{13}H_{16}N_2O$, m/e 216.1261).

Anal. Calcd for $\bar{C}_{13}H_{16}N_2O$: C, 72.19; H, 7.46; N, 12.95. Found: C, 72.18; H, 7.49; N, 12.93.

8-[2-(Methylamino)ethyl]-1,2,3,4-tetrahydroquinoline (7). α -(Methylthio)- α -(1,2,3,4-tetrahydroquinolin-8-yl)-N-methylacetamide (4) (7.5 g, 30 mmol) was desulfurized with activated Raney nickel as described for the preparation of 5 (vide supra). The resulting undistilled product was dissolved in 200 mL of dry tetrahydrofuran, and 2.0 g (52.7 mmol) of lithium aluminum hydride was added. The reaction mixture was refluxed for 4 h and hydrolyzed by adding, in sequence (slowly), 2 mL of water, 2 mL of 15% sodium hydroxide solution, and 6 mL of water. The resulting mixture was filtered through a fritted glass funnel, and the filtrate was concentrated to give an oil. Distillation of the oil gave 4.07 g (21.4 mmol, 71%) of 7: bp 106–112 °C (0.2 mm); IR (neat) 3300 (w), 2920 (m), 2830 (m), 1595 (m), 1500 (m), 1450 (m), 1350 (w), 1315 (m), 1280 (m), 1190 (w), 1115 (w) cm⁻¹; ¹H NMR (CDCl₃) δ 1.70–2.17 (m, 2 H), 2.38 (s, 3 H), 2.54–2.98 (m, 6 H), 3.16–3.44 (m, 2 H), 6.33–7.02 (complex m, 3 H); ¹³C NMR (CDCl₃) δ 21.86 (t), 27.36 (t), 31.62 (t), 36.37 (q), 41.01 (t), 51.34 (t), 116.17 (d), 121.28 (s), 123.5 (s), 127.41 (d), 127.64 (d), 142.80 (s); exact mass, *m/e* 190.1476 (calcd for C₁₂H₁₈N₂, *m/e* 190.1470).

Anal. Calcd for $C_{12}H_{18}N_2$: C, 75.74; H, 9.53; N, 14.72. Found: C, 75.45; H, 9.54; N, 14.60.

1,2,3,4,9,10-Hexahydro-2-methyl-8*H*-pyrido[3,2,1-*jk*][1,3]benzodiazepine (8). To 0.95 g (5 mmol) of 8-[2-(methylamino)ethyl]-1,2,3,4-tetrahydroquinoline (7) in 20 mL of benzene was added 0.15 g (5 mmol) of paraformaldehyde, and the reaction mixture was refluxed overnight. Concentration of the reaction mixture gave an oil, which was distilled to give 0.92 g (4.55 mmol, 91%) of 8: bp 100-102 °C (0.2 mm); IR (neat) 2960 (m), 2840 (w), 1605 (w), 1500 (m), 1370 (w), 1320 (m), 1300 (m), 1260 (w), 1220 (w), 1200 (w), 1120 (m), 1080 (w), 1060 (w) cm⁻¹; ¹H NMR (CDCl₃) δ 1.56-2.05 (m, 2 H), 2.47 (s, 3 H), 2.58-3.00 (m, 6 H), 3.12-3.43 (m, 2 H), 3.90 (s, 2 H), 6.49-7.02 (complex m, 3 H); ¹³C NMR (CDCl₃) δ 21.29 (t), 27.84 (t), 32.59 (t), 39.60 (q), 53.03 (t, 2 C), 77.86 (t), 119.35 (d), 126.03 (s), 126.89 (d), 128.13 (d), 131.80 (s), 147,72 (s); exact mass, m/e 202.1471 (calcd for $C_{13}H_{18}N_2, m/e$ 202.1470).

Anal. Calcd for C₁₃H₁₈N₂: C, 77.18; H, 8.97; N, 13.85. Found: C, 77.10; H, 9.03; N, 13.99.

2-Methyl-3,4,9,10-tetrahydro-8H-pyrido[3,2,1-ik][1,3]benzodiazepin-1(2H)-one (9). To 0.95 g (5 mmol) of 8-[2-(methylamino)ethyl]-1,2,3,4-tetrahydroquinoline (7) in 50 mL of dry tetrahydrofuran was added 0.81 g (5 mmol) of 1,1'carbonyldiimidazole, and the reaction mixture was refluxed for 1 h. To this reaction mixture was added 2.25 g (20 mmol) of potassium tert-butoxide, and the reaction mixture was refluxed for 2 h. The reaction mixture was concentrated, and 100 mL of methylene chloride was added. The mixture was washed with water, dried over anhydrous sodium carbonate, filtered, and concentrated to give a solid. This solid was recrystallized from ether-hexane to give 0.61 g (2.8 mmol, 56%) of 9: mp 121-123 °C; IR (CDCl₃) 2940 (w), 1620 (ms), 1490 (m), 1440 (m), 1400 (w), 1350 (w), 1330 (w), 1310 (w), 1250 (w), 1180 (w), 1060 (w) cm⁻¹; ¹H NMR (CDCl₃) δ 1.72–2.13 (m, 2 H), 2.76 (s, 3 H), 2.63–3.12 (m, 4 H), 3.20–3.85 (m, 4 H), 6.99 (s, 3 H); ^{13}C NMR (CDCl₃) δ 22.53 (t), 26.67 (t), 30.34 (t), 38.31 (q), 44.51 (t), 56.61 (t), 123.81 (d), 124.95 (d), 127.70 (d), 129.46 (s), 133.57 (s), 140.76 (s), 158.44 (s); exact mass, m/e 216.1260 (calcd for C₁₃H₁₆N₂O, m/e 216.1261).

Anal. Calcd for $C_{13}H_{16}N_2O$: C, 72.19; H, 7.46; N, 12.95. Found: C, 72.16; H, 7.54; N, 12.93.

Acknowledgment. We are indebted to the National Institutes of Health for Grant GM-22346, which supported this investigation.

A Short Synthesis of Elasnin

Robert L. Shone,* James R. Deason, and Masateru Miyano

Department of Medicinal Chemistry, Searle Research and Development, Division of G.D. Searle & Co., Skokie, Illinois 60077

Received August 6, 1985

Elasnin (1) is a complex trisubstituted 4-hydroxy-2pyrone that has been isolated from culture broths of *Streptomyces norboritoensis*.^{1a,b} It is a specific inhibitor

^{(1) (}a) Ohno, H.; Saheki, T.; Awaya, J., Nakagawa, A.; Omura, S. J. Antibiot. 1978, 31, 116. (b) Omura, S.; Ohno, H.; Saheki, T.; Yoshida, M.; Nakagawa, A. Biochem. Biophys. Res. Commun. 1978, 83, 704. (c) Omura, S.; Nakagawa, A.; Ohno, H. J. Am. Chem. Soc. 1979, 101, 4386. (d) Nakagawa, A.; Ohno, H.; Miyano, K.; Omura, S. J. Org. Chem. 1980, 45, 3268.